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### [4 + 2] Cycloaddition Reactions Involving 2-Arylmethylidene-1-thiooxindan Intermediates and Antimicrobial Activity Evaluation of Some Products

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## [4 + 2] Cycloaddition Reactions Involving 2-Arylmethylidene-1-thiooxindan Intermediates and Antimicrobial Activity Evaluation of Some Products

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*2-(4-Fluorophenyl)methylidene-1-thiooxindane-, 2-(4-methoxyphenyl) methylidene-1-thiooxindane-, 2-(4-N,N-dimethylaminophenyl)methylidene-1-thiooxindane dimers 5a–c were prepared by the reaction of the corresponding  $\alpha,\beta$ -unsaturated ketones 3a–c with Lawesson's Reagent (LR) in refluxing benzene. When these dimers were refluxed with LR in xylene, the 1,2-thiaphospholene-2-sulfides 7a–c were obtained. On the other hand, the thermolysis of the dimers 5a–c in the presence of acrylamide or dichloromaleic anhydride gave the corresponding cycloadducts of Diels–Alder type 8a–c, and 10a–c, respectively. The 3-carbamoyl thiapyran derivatives 8a–c showed good antimicrobial activity.*

**Keywords** Antimicrobial activity; 2-arylmethylidene-1-thiooxindane dimers; Diels–Alder cycloadducts; Lawesson's reagent; 1,2-thiaphospholene-2-sulfides;  $\alpha,\beta$ -unsaturated ketones

## INTRODUCTION

The  $\alpha,\beta$ -unsaturated thiones are little known because of their instability in the monomeric form<sup>1–9</sup> and tendency to undergo [4 + 2] cycloaddition in which the thione itself may serve as a dienophile or a diene. However, Karakasa and Motoki<sup>10</sup> reported the preparation of some thiochalcone dimers and 2-arylbenzylidene-1-thiotetralone dimers via the reaction of the corresponding  $\alpha,\beta$ -unsaturated ketones with  $P_4S_{10}$ . Also, the thermolysis of thiochalcone dimers and 2-arylbenzylidene-1-thiotetralone dimers in the presence of acrylonitrile or acrylamide was reported.<sup>10</sup>

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On the other hand, the reaction of some chalcones, 2-arylbenzylidene-1-tetralones and 2,2-dialkyl-3-arylbenzylidene-4-chromanones with 2,4-bis(4-methoxyphenyl)-1,3,2,4-dithiaphosphetane-2,4-disulfide (Lawesson's reagent, LR) was reported.<sup>11,12</sup> Recently, some new chromenothiapyran derivatives have been reported.<sup>13,14</sup> In continuation of synthesis of some new thiapyran derivatives, it is intended to explore the reaction of 2-arylbenzylidene-1-indanones with LR in order not only to isolate the corresponding 2-arylbenzylidene-1-thiooxoindanone dimers but also, to evaluate their antimicrobial activities.

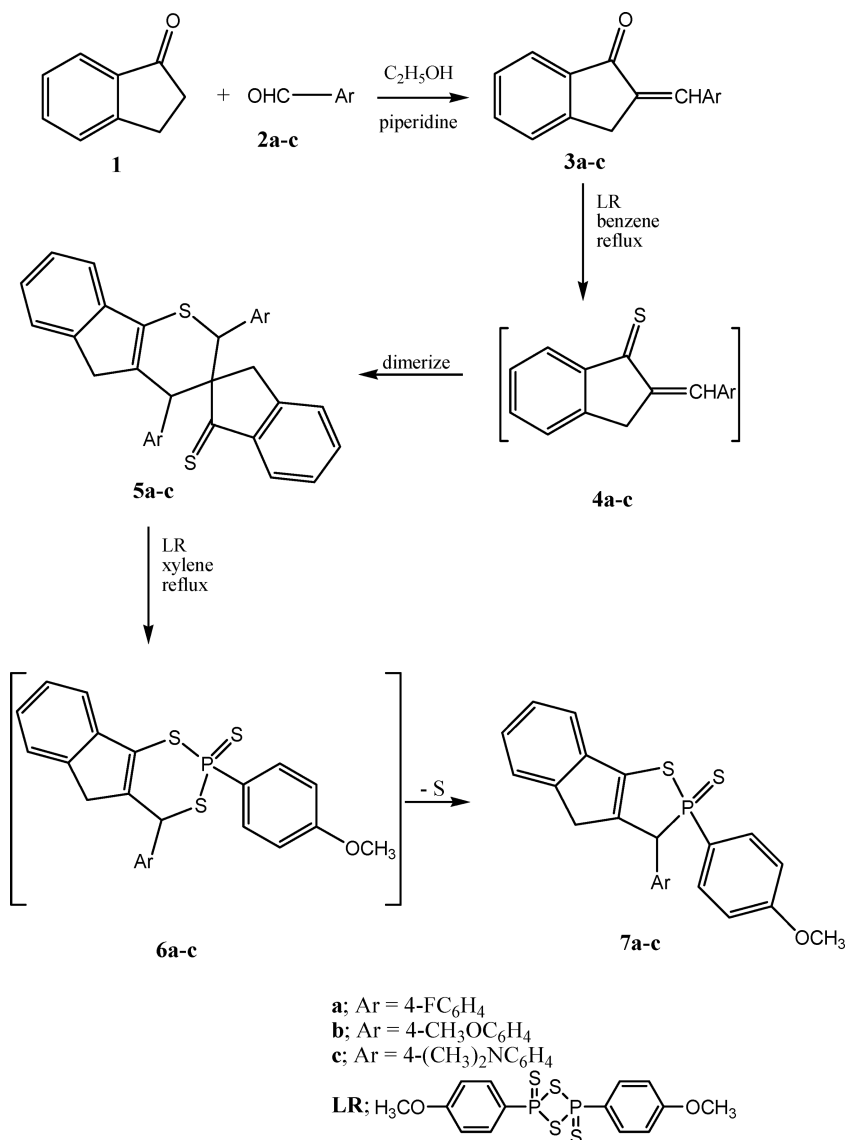
## RESULTS AND DISCUSSION

In the present work, three known 2-arylbenzylidene-1-indanones, namely, 2-(4-fluorobenzylidene)indan-1-one (**3a**), 2-(4-methoxybenzylidene)indan-1-one (**3b**), and 2-(4-*N,N*-dimethylaminobenzylidene)indan-1-one (**3c**) were prepared according to reference 15. Treatment of 2-(4-fluorobenzylidene)indan-1-one (**3a**) with LR in refluxing benzene for 3h gave the corresponding thiooxo dimer **5a** in moderate yield (Scheme 1). In this reaction, the straightforward thionation of the carbonyl group of **3** yielded the thiooxo monomer **4** which immediately underwent cyclization with another thiooxo monomer [2 + 4] to yield the thiooxo dimer **5**. The structure of **5a** was confirmed by analytical and spectral data. The <sup>1</sup>H NMR spectrum of **5a** showed one-proton singlets at 3.70 and 5.55 ppm; these were assigned to the C-4 and the C-2 protons in the 3,4-dihydro-2*H*-thiapyran ring, respectively. The mass spectrum of **5a** exhibited the molecular ion peak at 508 (*M*<sup>+</sup>, 2%) and a fragment from the thiooxo monomer **4a** (252, 100%, *4a*<sup>+</sup> – 2H) which would be formed by the cleavage of **5a**.

Similarly, 2-(4-methoxyphenyl)methylidene-1-thiooxoindan dimer (**5b**) and 2-(4-*N,N*-dimethylaminophenyl)methylidene-1-thiooxoindan dimer (**5c**) were obtained by the reaction of **3b**, and **3c** with LR, respectively (Scheme 1). The structures of **5b,c** are supported by analytical and spectral data (cf. Experimental).

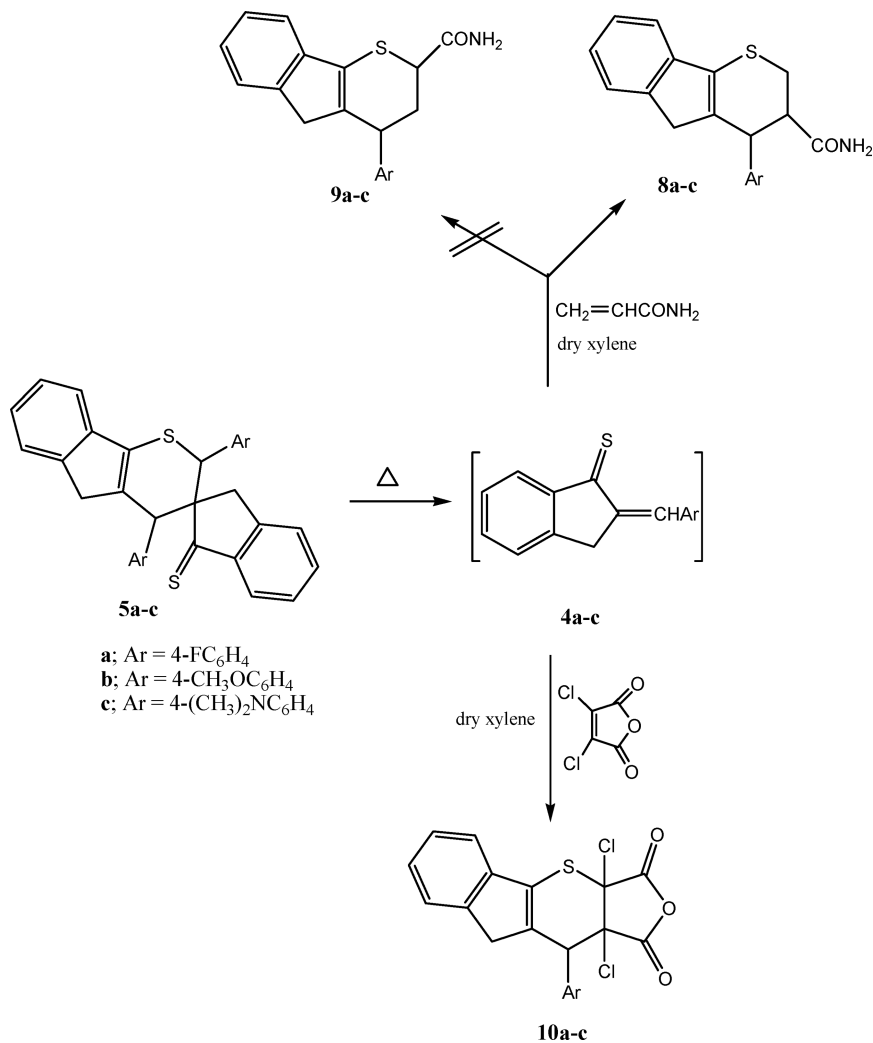
However, when the thiooxo dimers **5a–c** reacted with LR in refluxing xylene for 6 h, the phosphorus containing compounds **7a–c**, respectively, were formed (Scheme 1).

The structures of phosphorus containing compounds **7a–c** were confirmed by analytical and spectral data. The <sup>1</sup>H NMR spectrum of **7a** showed the doublet at 5.39 ppm with *J*<sub>PH</sub> = 15 Hz due to the presence of an adjacent CH moiety. Its mass spectrum showed the molecular ion peak at 424 (*M*<sup>+</sup>, 12%), and a base peak at 391 (*M*<sup>+</sup> – SH, 100%).



SCHEME 1

On the other hand, the thermolysis of  $\alpha,\beta$ -unsaturated thiooxo dimers **5a-c** in the presence of acrylamide in dry benzene for 1 h gave only adducts which were identified as **8a-c** based on <sup>1</sup>H NMR (cf. Scheme 2). In a competition reaction the unstable  $\alpha,\beta$ -unsaturated thiooxo monomer



SCHEME 2

**4** cyclized with an external dienophile rather than with itself. This reaction proved the presence of an unstable  $\alpha,\beta$ -unsaturated thioxo monomer as the intermediate. The  $^1\text{H}$  NMR of **8a** showed 4-CH at 5.54 ppm as a doublet with  $J = 3$  Hz and 3-CH at 4.79–4.89 as a multiplet. If derivative **9** would have formed, the 4-CH would appear as a triplet. Thus, according to the  $^1\text{H}$  NMR data the compounds **9a-c** could be excluded. The IR spectrum of compound **8a** showed bands at 3275, 2925 (NH<sub>2</sub>), and 1661 cm<sup>-1</sup> (C=O).

Products **8a–c** show clearly the  $^1\text{H}$  NMR signals of only one diastereomer. If the minor diastereomer is present, its concentration too small to be detected.

Similarly, the  $\alpha,\beta$ -unsaturated thioxo dimers **5a–c** were heated the presence of dichloromaleic anhydride in dry xylene for 3 h to give adducts **10a–c** (cf. Scheme 2). The IR spectrum of compound **10a** showed bands at 1773, 1723  $\text{cm}^{-1}$  [C(O)OC(O)]. Its  $^1\text{H}$  NMR spectrum revealed a signal at 5.55 (1H, s, CH). The mass spectrum of compound **10a** showed the molecular ion peak at 424 ( $\text{M}^+ 2\text{Cl}^{37}, 1$ ), 422 ( $\text{M}^+ \text{Cl}^{37,35}, 9$ ), and 420 ( $\text{M}^+ 2\text{Cl}^{35}, 50$ ).

## ANTIMICROBIAL ACTIVITY

The in vitro antimicrobial activity of the new synthesized thiapyran derivatives was investigated against several pathogenic representative Gram-negative bacteria, Gram-positive bacteria, fungi, and yeast. All microorganisms used were obtained from the culture collection of the Department of Natural and Microbial Products, National Research Centre, Dokki, Cairo, Egypt.

### Medium<sup>16–18</sup>

The cap-assay method containing (g/L): peptone 6, yeast extract 3, meat extract 1.5, glucose 1, and agar 20 was used. The medium was sterilized and divided while hot (50–60°C) in 15-mL portions among sterile 9-cm diameter Petri dishes. One mL of the spore suspension of each microorganism was spread all over the surface of the cold solid medium placed in the Petri dish.

### Method

0.5 g of each of the tested compounds was dissolved in 5 mL of *N,N*-dimethylformamide, an amount of 0.1 mL of test solution was placed on a 9-mm diameter Whatman paper disc, and the solvent was left to evaporate. These saturated discs were placed carefully on the surface of the incubated solid medium; each Petri dish contains at least 3 discs. The Petri dishes were incubated at 5°C overnight, then examined. The results were then recorded by measuring the inhibition zone diameters.

**TABLE I** The Antimicrobial Activity of Some Newly Synthesized Compounds

Tested Compounds and Standards	Inhibition zone (mm)			
	Microorganism			
	Bacteria		Fungi <i>Aspergillus Niger</i>	Yeast <i>Candida Albicans</i>
	Gram-negative	Gram-positive		
	<i>Escherichia Coli</i>	<i>Bacillus Subtilis</i>		
Streptomycin	+++	+++	+	+++
Fusidic Acid	—	—	+++	+++
<b>5a</b>	—	—	—	++
<b>5b</b>	—	—	—	+
<b>5c</b>	—	—	—	+
<b>8a</b>	+	+	—	+++
<b>8b</b>	+	+	—	++
<b>8c</b>	+	+	—	++
<b>10a</b>	—	—	—	++
<b>10b</b>	—	—	—	+
<b>10c</b>	—	—	—	+

+++ Highly sensitive (21–25 mm); ++ Fairly sensitive (16–20 mm); + Slightly sensitive (15–10 mm); —Not sensitive.

## RESULTS

The antimicrobial activity of the tested compounds (**5a–c**, **8a–c**, and **10a–c**) was evaluated by measuring the zone diameters and their results were compared with those of well-known drugs as shown in Table I. 3-Carbamoyl thiapyran derivatives **8a–c** showed good antimicrobial activity. However, 3-carbamoyl-4-(4-fluorophenyl)-2,3,4-trihydroindeno[1,2-*e*]thiapyran (**8a**) demonstrated inhibitory activity more than **8b** and **8c**.

## EXPERIMENTAL

Melting points are uncorrected and recorded on a digital Electrothermal IA 9000 SERIES melting point apparatus (Electrothermal, Essex, U.K.). Microanalyses were performed with all final compounds on Elementar-Vario EL, Microanalytical Unit, Central Services Laboratory, National Research Centre, Cairo, Egypt. The  $^1\text{H}$  NMR spectra were taken for samples in  $\text{CDCl}_3$  as solvent (unless otherwise mentioned) with Jeol EX-270 and Jeol EX-500 MHz NMR spectrometers, Central Services Laboratory, National Research Centre, Cairo, Egypt.

Chemical shifts are quoted in  $\delta$  and were referenced to that of the solvents. Splitting patterns were designated as follow: s singlet; d doublet; t triplet; m multiplet. Mass spectra were recorded on Shimadzu GCMS-QP 1000 EX EI (70 eV) spectrometers (Micro-Analytical Center of Cairo University). IR spectra were obtained with Brucker-Vector 22 for KBr wafers (Micro-analytical Center of Cairo University). Compounds **3a-c**<sup>15</sup> were prepared according to the literature procedure.

### Reaction of Arylbenzylidene-1-indanone Derivatives **3** with Lawesson's Reagent (LR)

A mixture of **3** (2 mmol) and LR (0.4 g, 1.1 mmol) in dry benzene (20 mL) was refluxed for 3 h. After cooling, the reaction mixture was filtered off and the filtrate was evaporated. The residue was chromatographed on silica gel (Fluka 60, particle size 0.06–0.20 mm) using diethyl ether:petroleum ether 40–60°C 1:10 (v/v) as an eluent. The solvent was evaporated and the residue was recrystallized from ethanol to give thiapyran derivatives **5**.

#### **2,4-Di(4-fluorophenyl)spiroindan-2',3-inden[1,2-b]-2,4-dihydrothiapyran-1'-thione (5a)**

From **3a** (0.46 g). Green crystals, m.p. 223–226°C, yield 55% (0.5 g). IR:  $\nu$  = 2915, 1690, 1605, 1510, 1460, 1250, 1171, 1031, 835, 759, 715  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR:  $\delta$  = 3.44 (2H, s,  $\text{CH}_2$ ), 3.52 (2H, s,  $\text{CH}_2$ ), 3.70 (1H, s, CH), 5.55 (1H, s, CH), 6.97–7.57 (16H, m, Ar-H). MS (EI):  $m/z$  (%) = 508 ( $\text{M}^+$ , 2), 475 (24), 424 (2), 394 (5), 367 (15), 254 (29), 252 (100), 219 (9), 139 (43), 126 (19), 115 (15). Anal. calcd. for  $\text{C}_{32}\text{H}_{22}\text{F}_2\text{S}_2$  (508.62): C, 75.56; H, 4.36; S, 12.60%. found: C, 75.34; H, 4.22; S, 12.29.

#### **2,4-Di(4-methoxyphenyl)spiroindan-2',3-inden[1,2-b]-2,4-dihydrothiapyran-1'-thione (5b)**

From **3b** (0.50 g). Green crystals, m.p. 109–111°C, yield 57% (0.57 g). IR:  $\nu$  = 2923, 1701, 1607, 1509, 1459, 1249, 1176, 1033, 833, 758, 717  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR:  $\delta$  = 3.43 (2H, s,  $\text{CH}_2$ ), 3.53 (2H, s,  $\text{CH}_2$ ), 3.70 (1H, s, CH), 3.85 (6H, s,  $\text{OCH}_3$ ), 5.75 (1H, s, CH), 6.50–7.35 (16H, m, Ar-H). MS (EI):  $m/z$  (%) = 266 ( $1/2\text{M}^+$ , 74), 265 (100), 251 (29), 235 (53), 221 (22), 189 (20), 151 (44), 121 (25), 110 (8). Anal. calcd. for  $\text{C}_{34}\text{H}_{28}\text{O}_2\text{S}_2$  (532.68): C, 76.65; H, 5.29; S, 12.03%. found: C, 76.34; H, 5.12; S, 11.79.

#### **2,4-Di(4-N,N-dimethylaminophenyl)spiroindan-2',3-inden[1,2-b]-2,4-dihydrothiapyran-1'-thione (5c)**

From **3c** (0.52 g). Dark green crystals, m.p. 223–226°C, yield 50% (0.55 g). IR:  $\nu$  = 2918, 1610, 1586, 1569, 1521, 1465, 1323, 1265, 1162,



1094, 1029, 947, 814, 758, 718  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR:  $\delta$  = 2.84 (12H, s, 2  $\text{N}(\text{CH}_3)_2$ ), 3.43 (2H, s,  $\text{CH}_2$ ), 3.50 (2H, s,  $\text{CH}_2$ ), 3.71 (1H, s, CH), 5.65 (1H, s, CH), 6.70–7.85 (16H, m, Ar-H). MS (EI):  $m/z$  (%) = 279 (1/2 $\text{M}^+$ , 61), 278 (100), 263 (11), 235 (21), 221 (2), 189 (3), 164 (49), 148 (17), 139 (25), 132 (14), 121 (17), 115 (14). Anal. calcd. for  $\text{C}_{36}\text{H}_{34}\text{N}_2\text{S}_2$  (558.77): C, 77.37; H, 6.13; N, 5.01%. found: C, 77.04; H, 5.98; N, 4.79.

### Heating of Thioxo Dimers 5 with Lawesson's Reagent (LR)

A mixture of **5** (2 mmol) and LR (0.8 g, 2.1 mmol) in dry xylene (20 mL) was refluxed for 6 h. After cooling, the reaction mixture was filtered off and the filtrate was evaporated. The residue was chromatographed on silica gel (Fluka 60, particle size 0.06–0.20 mm) using diethyl ether:petroleum ether 40–60°C 1:10 (v/v) as an eluent. The solvent was evaporated and the residue was recrystallized from ethanol to give phosphorus compounds **7**.

#### **3-(4-Fluorophenyl)-2-(4-methoxyphenyl)indeno[1,2-d]-3H-1,2-thiaphospholene-2-sulfide (7a)**

From **5a** (1.0 g). Colorless crystals, m.p. 201–204°C, yield 60% (0.48 g).  $^1\text{H}$  NMR:  $\delta$  = 3.51 (2H, s,  $\text{CH}_2$ ), 3.79 (3H, s,  $\text{OCH}_3$ ), 5.39 (1H, d,  $J_{\text{PH}}$  = 15Hz, 3-CH), 6.77–7.67 (12H, m, Ar-H). MS (EI):  $m/z$  (%) = 424 ( $\text{M}^+$ , 12), 391 (100), 345 (50), 254 (29), 222 (19), 139 (33), 126 (29), 115 (16). Anal. calcd. for  $\text{C}_{23}\text{H}_{18}\text{FOPS}_2$  (424.48): C, 65.07; H, 4.27; S, 15.10%. found: C, 64.81; H, 4.07; S, 14.93.

#### **2,3-Di(4-methoxyphenyl)indeno[1,2-d]-3H-1,2-thiaphospholene-2-sulfide (7b)**

From **5b** (1.0 g). Colorless crystals, m.p. 91–94°C, yield 63% (0.54 g).  $^1\text{H}$  NMR:  $\delta$  = 3.50 (2H, s,  $\text{CH}_2$ ), 3.68 (3H, s,  $\text{OCH}_3$ ), 3.80 (3H, s,  $\text{OCH}_3$ ), 5.39 (1H, d,  $J_{\text{PH}}$  = 15Hz, 3-CH), 6.77–7.67 (12H, m, Ar-H). MS (EI):  $m/z$  (%) = 436 ( $\text{M}^+$ , 12), 404 (100), 379 (50), 266 (19), 222 (39), 139 (23), 126 (19), 115 (15). Anal. calcd. for  $\text{C}_{24}\text{H}_{21}\text{O}_2\text{PS}_2$  (436.52): C, 65.03; H, 4.85; S, 14.68%. found: C, 64.85; H, 4.57; S, 14.43.

#### **3-(4-N,N-Dimethylphenyl)-2-(4-methoxyphenyl)indeno[1,2-d]-3H-1,2-thiaphospholene-2-sulfide (7c)**

From **5c** (1.1 g). Colorless crystals, m.p. 198–201°C, yield 45% (0.20 g).  $^1\text{H}$  NMR:  $\delta$  = 2.86 (6H, s,  $\text{N}(\text{CH}_3)_2$ ), 3.52 (2H, s,  $\text{CH}_2$ ), 3.81 (3H, s,  $\text{OCH}_3$ ), 5.40 (1H, d,  $J_{\text{PH}}$  = 15Hz, 3-CH), 6.97–7.77 (12H, m, Ar-H). MS (EI):  $m/z$  (%) = 449 ( $\text{M}^+$ , 10), 416 (100), 279 (50), 221 (20), 139 (33), 126 (29). Anal. calcd. for  $\text{C}_{25}\text{H}_{24}\text{NOPS}_2$  (449.56): C, 66.78; H, 5.38; N, 3.11; S, 14.26%. found: C, 66.46; H, 5.19; N, 2.98; S, 14.03.

## The Thermolysis of $\alpha,\beta$ -unsaturated thioxo dimers 5 in the Presence of Acrylamide

A solution of  $\alpha,\beta$ -unsaturated thioxo dimers **5** (1 mmol) and acrylamide (0.15 g, 2.1 mmol) in dry benzene (10 mL) was refluxed for 1 h. The solvent was evaporated and the residue was chromatographed on silica gel (Fluka 60, particle size 0.06–0.20 mm) using petroleum ether 40–60°C as an eluent. The solvent was evaporated and the residue was recrystallized from ethanol to give the adduct **8**.

### 3-Carbamoyl-4-(4-fluorophenyl)-2,3,4-trihydroindeno[1,2-*e*]thiapyran (**8a**)

From **5a** (0.5 g). Colorless crystals, m.p. 108–110°C, yield 30% (0.1 g). IR:  $\nu = 3275, 2925, 1661, 1625, 1515, 1464, 1250, 1173, 1031, 833, 761, 715\text{ cm}^{-1}$ .  $^1\text{H NMR}$ :  $\delta = 3.50$  (2H, s,  $\text{CH}_2$ ), 3.85 (2H, d,  $J = 3\text{ Hz}$ ,  $\text{CH}_2$ ), 4.79–4.89 (1H, m, CH), 5.45 (1H, d,  $J = 3\text{ Hz}$ , CH), 5.85 (2H, br. s,  $\text{NH}_2$ ), 6.95–7.35 (8H, m, Ar-H). MS (EI):  $m/z$  (%) = 325 ( $\text{M}^+$ , 25), 254 (19), 235 (100), 189 (9), 118 (43). Anal. calcd. for  $\text{C}_{19}\text{H}_{16}\text{FNOS}$  (325.38): C, 70.13; H, 4.95; N, 4.30; S, 9.85%. found: C, 69.84; H, 4.82; N, 4.18; S, 9.64.

### 3-Carbamoyl-4-(4-methoxyphenyl)-2,3,4-trihydroindeno[1,2-*e*]thiapyran (**8b**)

From **5b** (0.5 g). Colorless crystals, m.p. 65–68°C, yield 38% (0.12 g). IR:  $\nu = 3275, 2926, 1665, 1627, 1515, 1462, 1250, 1171, 1031, 835, 763, 717\text{ cm}^{-1}$ .  $^1\text{H NMR}$ :  $\delta = 3.50$  (2H, s,  $\text{CH}_2$ ), 3.83 (3H, s,  $\text{OCH}_3$ ), 3.86 (2H, d,  $J = 3\text{ Hz}$ ,  $\text{CH}_2$ ), 4.77–4.85 (1H, m, CH), 5.45 (1H, d,  $J = 3\text{ Hz}$ , CH), 5.84 (2H, br. s,  $\text{NH}_2$ ), 6.95–7.55 (8H, m, Ar-H). MS (EI):  $m/z$  (%) = 337 ( $\text{M}^+$ , 45), 266 (39), 235 (100), 189 (19), 118 (23). Anal. calcd. for  $\text{C}_{20}\text{H}_{19}\text{NO}_2\text{S}$  (337.41): C, 71.89; H, 5.67; N, 4.14; S, 9.50%. found: C, 71.54; H, 5.82; N, 3.88; S, 9.24.

### 3-Carbamoyl-4-(4-*N,N*-dimethylaminophenyl)-2,3,4-trihydroindeno[1,2-*e*]thiapyran (**8c**)

From **5c** (0.5 g). Colorless crystals, m.p. 112–115°C, yield 32% (0.11 g). IR:  $\nu = 3275, 2926, 1660, 1625, 1517, 1460, 1250, 1171, 1031, 837, 763, 715\text{ cm}^{-1}$ .  $^1\text{H NMR}$ :  $\delta = 2.85$  (6H, s,  $\text{N}(\text{CH}_3)_2$ ), 3.51 (2H, s,  $\text{CH}_2$ ), 3.85 (2H, d,  $J = 3\text{ Hz}$ ,  $\text{CH}_2$ ), 4.78–4.87 (1H, m, CH), 5.45 (1H, d,  $J = 3\text{ Hz}$ , CH), 5.85 (2H, br. s,  $\text{NH}_2$ ), 6.95–7.55 (8H, m, Ar-H). MS (EI):  $m/z$  (%) = 350 ( $\text{M}^+$ , 15), 234 (100), 189 (29), 118 (14). Anal. calcd. for  $\text{C}_{21}\text{H}_{22}\text{N}_2\text{OS}$  (350.45): C, 71.96; H, 6.32; N, 7.99; S, 9.14%. found: C, 71.64; H, 6.22; N, 7.68; S, 8.84.

## The Thermolysis of $\alpha,\beta$ -unsaturated Thioxo Dimers 5 in the Presence of Dichloromaleic Anhydride

A solution of  $\alpha,\beta$ -unsaturated thioxo dimers **5** (1 mmol) and dichloromaleic anhydride (3.32 g, 2.1 mmol) in dry xylene (10 mL) was refluxed for 3 h. The solvent was evaporated and the residue was chromatographed on silica gel (Fluka 60, particle size 0.06–0.20 mm) using petroleum ether 40–60°C as an eluent. The solvent was evaporated and the residue was recrystallized from ethyl acetate to give the adduct **10**.

### **3a,10a-Dichloro-10-(4-fluorophenyl)-3a,9,10,10a-tetrahydroindeno[1,2-e]thiapyran[2,3-d]furan-1,3-dione (10a)**

From **5a** (0.5 g). Colorless crystals, m.p. 118–121°C, yield 25% (0.2 g). IR:  $\nu$  = 2925, 1773, 1723, 1606, 1511, 1463, 1250, 1177, 1032, 916, 834, 757  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR:  $\delta$  = 3.50 (2H, s,  $\text{CH}_2$ ), 5.55 (1H, s, CH), 6.95–7.55 (8H, m, Ar-H). MS (EI):  $m/z$  (%) = 424 ( $\text{M}^+$   $2\text{Cl}^{37}$ , 1), 422 ( $\text{M}^+$   $\text{Cl}^{37,35}$ , 9), 420 ( $\text{M}^+$   $2\text{Cl}^{35}$ , 50), 350 (19), 306 (9), 235 (100), 189 (19), 152 (15). Anal. calcd. for  $\text{C}_{20}\text{H}_{11}\text{Cl}_2\text{FO}_3\text{S}$  (421.26): C, 57.01; H, 2.63; S, 7.61%. found: C, 56.88; H, 2.55; S, 7.34.

### **3a,10a-Dichloro-10-(4-methoxyphenyl)-3a,9,10,10a-tetrahydroindeno[1,2-e]thiapyran[2,3-d]furan-1,3-dione (10b)**

From **5b** (0.5 g). Colorless crystals, m.p. 78–81°C, yield 27% (0.2 g). IR:  $\nu$  = 2924, 1771, 1723, 1604, 1509, 1461, 1248, 1175, 1030, 915, 832, 755  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR:  $\delta$  = 3.50 (2H, s,  $\text{CH}_2$ ), 3.84 (3H, s,  $\text{OCH}_3$ ), 5.53 (1H, s, CH), 6.97–7.58 (8H, m, Ar-H). MS (EI):  $m/z$  (%) = 436 ( $\text{M}^+$   $2\text{Cl}^{37}$ , 2), 434 ( $\text{M}^+$   $\text{Cl}^{37,35}$ , 18), 432 ( $\text{M}^+$   $2\text{Cl}^{35}$ , 50), 399 (3), 397 (15), 384 (100), 369 (20), 363 (4), 265 (49), 250 (12), 235 (20), 221 (8), 189 (8), 151 (19). Anal. calcd. for  $\text{C}_{21}\text{H}_{14}\text{Cl}_2\text{O}_4\text{S}$  (433.29): C, 58.20; H, 3.25; S, 7.39%. found: C, 57.98; H, 3.05; S, 7.14.

### **3a,10a-Dichloro-10-(4-*N,N*-dimethylaminophenyl)-3a,9,10,10a-tetrahydroindeno[1,2-e]thiapyran[2,3-d]furan-1,3-dione (10c)**

From **5c** (0.5 g). Colorless crystals, m.p. 123–125°C, yield 25% (0.2 g). IR:  $\nu$  = 2924, 1775, 1725, 1606, 1507, 1465, 1248, 1175, 1030, 917, 832, 757  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR:  $\delta$  = 2.86 (6H, s,  $\text{N}(\text{CH}_3)_2$ ), 3.52 (2H, s,  $\text{CH}_2$ ), 5.51 (1H, s, CH), 6.96–7.57 (8H, m, Ar-H). MS (EI):  $m/z$  (%) = 449 ( $\text{M}^+$   $2\text{Cl}^{37}$ , 2), 447 ( $\text{M}^+$   $\text{Cl}^{37,35}$ , 18), 445 ( $\text{M}^+$   $2\text{Cl}^{35}$ , 50), 412 (3), 410 (16), 375 (10), 360 (5), 331 (39), 250 (15), 235 (21), 221 (7), 189 (10), 151 (20). Anal. calcd. for  $\text{C}_{22}\text{H}_{17}\text{Cl}_2\text{NO}_3\text{S}$  (446.33): C, 59.19; H, 3.84; N, 3.13%. found: C, 58.88; H, 3.65; N, 2.79.

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